

## Refinement

Refinement on  $F^2$  $R(F) = 0.030$  $wR(F^2) = 0.080$  $S = 1.13$ 

567 reflections

115 parameters

All H-atom parameters  
refined

$$w = 1/[\sigma^2(F_o^2) + 0.0425F_o^2]$$

$$(\Delta/\sigma)_{\max} = 0.03$$

$$\Delta\rho_{\max} = 0.12 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.14 \text{ e } \text{\AA}^{-3}$$

Extinction correction: none

Atomic scattering factors  
from SHELXL93  
(Sheldrick, 1993)*Acta Cryst.* (1995). C51, 1304–1306

## 5-Nitrofuran-2-aldoxime

TOMASZ A. OLSZAK

*Department of Crystallography, University of Łódź,  
Pomorska 149/153, PL-92236 Łódź, Poland*OSWALD M. PEETERS, NORBERT M. BLATON AND  
CAMIEL J. DE RANTER*Laboratorium voor Analytische Chemie en  
Medicinale Fysicochemie, Faculteit Farmaceutische  
Wetenschappen, Katholieke Universiteit Leuven,  
Van Evenstraat 4, B-3000 Leuven, Belgium*

(Received 26 May 1994; accepted 15 July 1994)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

$$U_{\text{eq}} = (1/3)\sum_i\sum_j U_{ij}a_i^*a_j^*a_i\cdot a_j$$

	x	y	z	$U_{\text{eq}}$
O1	0.8168 (2)	0.4839	0.2696 (2)	0.038 (1)
O2	0.8653 (4)	-0.0586 (8)	0.0535 (2)	0.064 (1)
O3	1.0886 (3)	0.1174 (8)	0.1948 (2)	0.063 (1)
O4	0.7920 (3)	1.1446 (8)	0.5261 (2)	0.045 (1)
N1	0.9158 (4)	0.1098 (9)	0.1390 (2)	0.045 (1)
N2	0.8180 (3)	0.9223 (8)	0.4431 (2)	0.037 (1)
C1	0.6377 (4)	0.6235 (10)	0.2922 (2)	0.038 (1)
C2	0.4776 (4)	0.5279 (9)	0.2132 (3)	0.046 (1)
C3	0.5575 (4)	0.3163 (10)	0.1381 (3)	0.046 (1)
C4	0.7612 (4)	0.2999 (9)	0.1757 (2)	0.039 (1)
C5	0.6468 (4)	0.8476 (9)	0.3854 (3)	0.039 (1)

Table 2. Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

O1—C4	1.347 (4)	N2—C5	1.272 (3)
O1—C1	1.366 (4)	C1—C2	1.358 (4)
O2—N1	1.230 (4)	C1—C5	1.441 (5)
O3—N1	1.224 (3)	C2—C3	1.397 (5)
O4—N2	1.382 (3)	C3—C4	1.348 (4)
N1—C4	1.413 (4)		
C4—O1—C1	104.8 (2)	O1—C1—C5	118.6 (2)
O3—N1—O2	124.0 (3)	C1—C2—C3	106.9 (3)
O3—N1—C4	118.9 (3)	C4—C3—C2	105.4 (3)
O2—N1—C4	117.0 (2)	O1—C4—C3	112.5 (3)
C5—N2—O4	111.8 (2)	O1—C4—N1	116.9 (2)
C2—C1—O1	110.4 (3)	C3—C4—N1	130.4 (3)
C2—C1—C5	130.9 (3)	N2—C5—C1	121.3 (3)

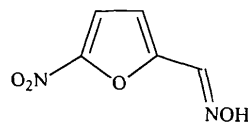
Computations were performed using the SHELXL93 package (Sheldrick, 1993).

One of us (VKB) is thankful to the Am. Crystallogr. Assoc./USNCCr Fund for financial support.

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: AS1123). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

## References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–S19.
- Chemla, D. S. & Zyss, J. (1987). Editors. *Nonlinear Optical Properties of Organic Molecules and Crystals*, Vols 1 and 2. New York: Academic Press.
- Matsuoka, M., Furukawa, M., Takao, M., Kitao, T., Hamada, M. & Nakatsu, K. (1991). *Chem. Lett.* pp. 289–292.
- Olszak, T. A., Peeters, O. M., Blaton, N. M. & De Ranter, C. J. (1995). *Acta Cryst.* C51, 1304–1306.
- Sheldrick, G. M. (1993). SHELXL93. Program for the Refinement of Crystal Structures. Univ. of Göttingen, Germany.



A perspective view showing the atomic numbering scheme and hydrogen bonds is given in Fig. 1. The crystal contains well ordered molecules of 5-nitro-2-furaldehyde semicarbazone (Olszak, Peeters, Blaton & De Ranter, 1994). The bond lengths are similar to within 0.025  $\text{\AA}$ . The title compound has a *cis* conformation while the one cited is *trans* with respect to the double bond of the side chain (C21=N22). A comparison with the bond lengths given by Matsuoka *et al.* (1991) for the title compound shows they are the same to within 0.019  $\text{\AA}$  for bonds involving non-H atoms.

The planar C3=C2—C21=N22 group of atoms does not show conjugation between the double bonds. A comparison of the bond lengths with those given by Allen *et al.* (1987) and Burke-Laing & Laing (1976)

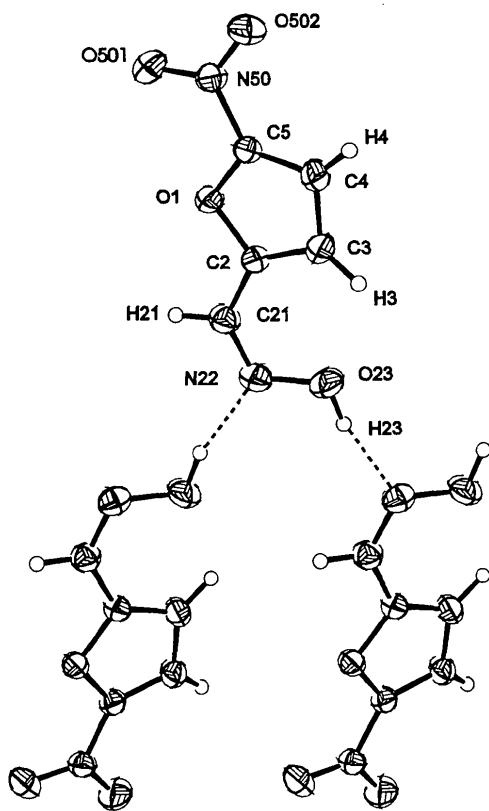


Fig. 1. The atomic numbering scheme and hydrogen bonds. Displacement ellipsoids are plotted at the 40% probability level.

shows a bond order of two for  $C3=C2$  and  $C21=N22$ , and one for  $C2-C21$ .

The furan ring is planar to within experimental error. The plane of the nitro substituent crosses the five-membered ring plane at an angle of  $0.85(5)^\circ$ . The side chain is also planar and its least-squares plane makes a dihedral angle of  $6.52(5)^\circ$  with the furan ring. The molecule, except for the  $C21$  and  $O23$  atoms, is almost planar (to within  $0.06 \text{ \AA}$ ). The slight deviation from planarity of the side chain [ $C21$  and  $O23$  are  $0.082(1)$  and  $0.114(1) \text{ \AA}$ , respectively, out of the plane] is caused by the *cis* conformation [ $O23 \cdots H3$  distance is  $2.40(2) \text{ \AA}$ ] and by hydrogen bonding.

The molecules are linked by intermolecular hydrogen bonds  $O23-H23 \cdots N22$ , where  $N22$  belongs to the molecule related by  $x+\frac{1}{2}, -y-\frac{1}{2}, -z$ . The  $O23 \cdots N22$  and  $H23 \cdots N22$  distances are  $2.820(2)$  and  $1.97(2) \text{ \AA}$ , respectively, and the angle  $O23-H23 \cdots N22$  is  $170(2)^\circ$ . Thus the crystal structure is built from chains of molecules linked by hydrogen bonds around the  $2_1$  screw axis; these chains run in the  $[100]$  direction.

## Experimental

### Crystal data

$C_5H_4N_2O_4$   
 $M_r = 156.10$

Mo  $K\alpha$  radiation  
 $\lambda = 0.71069 \text{ \AA}$

### Orthorhombic

$P2_12_12_1$

$a = 5.3330(4) \text{ \AA}$

$b = 5.6330(6) \text{ \AA}$

$c = 21.203(3) \text{ \AA}$

$V = 637.0(1) \text{ \AA}^3$

$Z = 4$

$D_x = 1.6278 \text{ Mg m}^{-3}$

### Cell parameters from 50 reflections

$\theta = 21.50\text{--}28.44^\circ$

$\mu = 0.1347 \text{ mm}^{-1}$

$T = 293 \text{ K}$

Block

$0.20 \times 0.10 \times 0.10 \text{ mm}$

Colourless

### Data collection

Stoe Stadi-4 four-circle diffractometer

$\omega$  scans

Absorption correction: none

3176 measured reflections

1458 independent reflections

1244 observed reflections

$[I > 2\sigma(I)]$

$R_{\text{int}} = 0.0123$

$\theta_{\text{max}} = 27.50^\circ$

$h = -6 \rightarrow 6$

$k = 0 \rightarrow 7$

$l = -27 \rightarrow 27$

3 standard reflections

frequency: 60 min

intensity decay:

$< 3.0\%$

### Refinement

Refinement on  $F^2$

$R(F) = 0.0255$

$wR(F^2) = 0.0685$

$S = 1.101$

1451 reflections

116 parameters

All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0497P)^2]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.002$

$\Delta\rho_{\text{max}} = 0.189 \text{ e \AA}^{-3}$

$\Delta\rho_{\text{min}} = -0.146 \text{ e \AA}^{-3}$

Extinction correction: none

Atomic scattering factors

from *International Tables for Crystallography* (1992, Vol. C, Tables 4.2.6.8 and 6.1.1.4)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

$$U_{\text{eq}} = (1/3)\sum_i \sum_j U_{ij} a_i^* a_j^*$$

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{eq}}$
O1	0.0306 (2)	0.4555 (1)	0.11689 (4)	0.0379 (2)
C2	0.1844 (2)	0.2700 (2)	0.10019 (5)	0.0356 (3)
C21	0.0793 (2)	0.1147 (2)	0.05255 (6)	0.0428 (3)
N22	0.1849 (2)	-0.0599 (2)	0.02526 (5)	0.0462 (3)
O23	0.4288 (2)	-0.0945 (2)	0.04624 (5)	0.0555 (3)
C3	0.3979 (2)	0.2736 (3)	0.13548 (6)	0.0434 (3)
C4	0.3785 (2)	0.4712 (2)	0.17634 (6)	0.0434 (4)
C5	0.1566 (2)	0.5710 (2)	0.16251 (5)	0.0351 (3)
N50	0.0398 (2)	0.7762 (2)	0.18854 (5)	0.0409 (3)
O501	-0.1643 (2)	0.8378 (2)	0.16800 (5)	0.0585 (4)
O502	0.1553 (2)	0.8787 (2)	0.23016 (5)	0.0586 (3)

Table 2. Selected geometric parameters ( $\text{\AA}, ^\circ$ )

O1—C2	1.374 (1)	O23—H23	0.86 (2)
O1—C5	1.346 (1)	C3—C4	1.415 (1)
C2—C21	1.449 (1)	C4—C5	1.342 (1)
C2—C3	1.363 (1)	C5—N50	1.424 (1)
C21—N22	1.272 (1)	N50—O501	1.222 (1)
N22—O23	1.388 (1)	N50—O502	1.221 (1)
C2—O1—C5	104.8 (1)	C3—C4—C5	105.1 (1)
O1—C2—C3	110.2 (1)	O1—C5—C4	113.2 (1)
O1—C2—C21	114.1 (1)	C4—C5—N50	129.8 (1)
C21—C2—C3	135.6 (1)	O1—C5—N50	116.9 (1)
C2—C21—N22	127.8 (1)	C5—N50—O502	116.3 (1)
C21—N22—O23	112.2 (1)	C5—N50—O501	118.8 (1)
N22—O23—H23	106 (1)	O501—N50—O502	124.9 (1)
C2—C3—C4	106.6 (1)		

Data collection: *DIF4* (Stoe & Cie, 1992a). Cell refinement: *DIF4*. Data reduction: *REDU4* (Stoe & Cie, 1992b). Pro-

gram(s) used to solve structure: *SHELXS86* (Sheldrick, 1985). Program(s) used to refine structure: *SHELXL93* (Sheldrick, 1993). Molecular graphics: *ORTEX2.1* (McArdle, 1994). Software used to prepare material for publication: *PARST* (Nardelli, 1983).

The authors thank Norwich Eaton Pharmaceuticals, Inc. (Norwich, New York) for providing the crystals. One of them (TAO) is indebted to the Research Council of the Katholieke Universiteit Leuven (Belgium) for a Senior Fellowship (No. F/92/32).

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: NA1116). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

## References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–S19.
- Burke-Laing, M. & Laing, M. (1976). *Acta Cryst. B32*, 3216–3224.
- McArdle, P. (1994). *J. Appl. Cryst.* **27**, 438–439.
- Matsuoka, M., Furukawa, M., Takao, M., Kitao, T., Hamada, M. & Nakatsu, K. (1991). *Chem. Lett.* pp. 289–292.
- Nardelli, M. (1983). *Comput. Chem.* **7**, 95–98.
- Olszak, T. A., Peeters, O. M., Blaton, N. M. & De Ranter, C. J. (1994). *Acta Cryst. C50*, 948–950.
- Sheldrick, G. M. (1985). *SHELXS86. Program for the Solution of Crystal Structures*. Univ. of Göttingen, Germany.
- Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. Univ. of Göttingen, Germany.
- Stoe & Cie (1992a). *DIF4. Diffractometer Control Program*. Version 7.0. Stoe & Cie, Darmstadt, Germany.
- Stoe & Cie (1992b). *REDU4. Data Reduction Program*. Version 7.03. Stoe & Cie, Darmstadt, Germany.

*Acta Cryst.* (1995). **C51**, 1306–1310

## Two Methyl-Substituted Carbapenem Antibiotic Precursors

ANGÈLE CHIARONI, CLAUDE RICHE, MIREILLE ADONIAS,  
JOSEFA ANAYA, STEFAN D. GÉRO AND  
CATHERINE TACHDJIAN

*Institut de Chimie des Substances Naturelles, CNRS,  
91198 Gif-sur-Yvette CEDEX, France*

(Received 5 October 1994; accepted 12 December 1994)

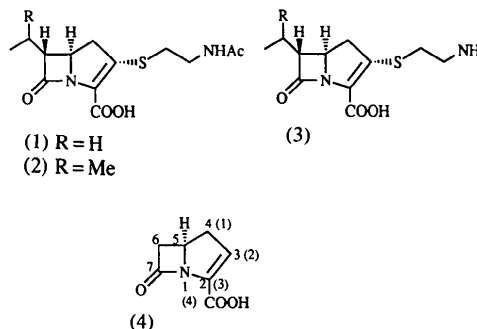
### Abstract

Carbapenem antibiotics are characterized by the presence of the 7-oxo-1-azabicyclo[3.2.0]hept-2-ene-2-carboxylic acid system. These new bicyclic  $\beta$ -lactam

antibiotics are rapidly degraded by dehydropeptidase-I. The introduction of a methyl group at the C1 position of the carbapenem skeleton improves the dehydropeptidase stability. The crystal structure determinations of two synthetic methyl-substituted carbapenem precursors, 4-benzyl-6-methoxy-3-propylsulfonyl-2-{2,2,2',2'-tetramethyl-[4,4'-bi([1,3]dioxolanyl)-5-yl]}-1-azabicyclo[3.2.0]heptane-7-one, C<sub>27</sub>H<sub>39</sub>NO<sub>8</sub>S, and 3-ethylthio-4-(2-furylmethyl)-6-methoxy-2-{2,2,2',2'-tetramethyl-[4,4'-bi([1,3]dioxolanyl)-5-yl]}-1-azabicyclo[3.2.0]heptane-7-one, C<sub>24</sub>H<sub>35</sub>NO<sub>7</sub>S, established their stereochemistry unambiguously. The absolute configuration was deduced from that of the chiral D-glucosamine auxiliary.

### Comment

Carbapenem antibiotics such as PS-5, (1) (Yamamoto *et al.*, 1980), PS-6, (2) (Ishikura, 1979), and thienamycin, (3) (Kahan *et al.*, 1979), comprise an interesting family of streptomycete metabolites characterized by the presence of the 7-oxo-1-azabicyclo[3.2.0]hept-2-ene-2-carboxylic acid system (4).



The potent antibacterial properties and the challenging chemical problems have made these new bicyclic  $\beta$ -lactams major synthetic objectives. A disadvantage of these compounds is that they are rapidly degraded in the kidney by dehydropeptidase-I (DHP-I) (Kahan *et al.*, 1979). The introduction of a methyl group into the 1-position of the carbapenem skeleton considerably improves the DHP stability (Neu, Novelli & Chin, 1989).

The synthesis of 1,2,3,6-tetrasubstituted carbapenem (6) from the readily available 1,3,4-trisubstituted azetidin-2-ones (5) (Barton *et al.*, 1990), in which the five-membered ring is formed by radical cyclization, has been reported recently (Anaya *et al.*, 1993). In continuing to explore the use of radical cyclization in the preparation of methyl-substituted carbapenem antibiotic precursors, we have synthesized compound (8) in 62% yield from the monocyclic  $\beta$ -lactam (7), which was obtained by Staudinger reaction using D-glucosamine as the chiral auxiliary (to be published). The X-ray struc-